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1 **Influences of AlN/Al nanoparticles on the creep properties of Elektron21 prepared by**  
2 **high shear dispersion technology**

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8 Abstract

9

10 Elektron21 (E21) and its composites with additions of 0.25 wt.%, 0.5 wt.% and 1 wt.% AlN/Al  
11 nanoparticles (NPs) were fabricated by a high shear dispersion technology (HSDT). Their creep  
12 properties were investigated over a stress range between 80-140 MPa at 240 °C. The grain size  
13 exhibits an obvious increase with the addition of AlN/Al NPs compared with monolithic E21  
14 alloy. Increasing the content of AlN/Al NPs leads to a pronounced improvement of creep  
15 resistance. Microstructural analysis shows that with the addition of 1% AlN/Al NPs in E21, the  
16 distribution of the intermetallics Mg<sub>3</sub>RE becomes much more homogeneous and their size is  
17 reduces. Such Mg<sub>3</sub>RE particles can prevent the dislocation slip more efficiently during creep.  
18 Besides these Mg<sub>3</sub>RE particles, the additional formation of Al<sub>2</sub>RE and Al<sub>2</sub>Zr<sub>3</sub> phases, which  
19 results from the reactions of AlN/Al NPs and the alloying elements Zr and REs, could act as  
20 thermal stable particles to improve the creep resistance. Finally, the remained AlN NPs without  
21 reactions are beneficial for the improvement of the creep resistance to some extent due to  
22 Orowan strengthening.

23 **1. Introduction**

24 Magnesium (Mg)-based alloys have attracted great interests in aerospace, transportation and  
25 manufacture industries because of their light weight and specific strength. However, their wider  
26 applications are restricted due to their relatively poor creep resistance at elevated temperatures,  
27 which made them unqualified to operate in automobile transmission cases or powertrain  
28 applications up to 175 °C [1]. In order to overcome this problem, ceramic particles, such as  
29 SiC [2], Al<sub>2</sub>O<sub>3</sub> [3] and AlN [4] with low density, low coefficient of thermal expansion and high  
30 thermal stability, were commonly incorporated into Mg-based alloys to enhance their creep

1 resistance. Especially AlN nanoparticles (NPs), with a hexagonal close packed (hcp) structure  
2 [5], were usually selected as a reinforcement in Mg matrix composites [6, 7].

3 The traditional preparation process, including powder metallurgy [8], high-energy ball  
4 milling [9] and stir casting [10] methods, either lack in safety or create insufficient energy to  
5 enable a uniform distribution of particles into the matrix. Ultrasonic treatment was considered  
6 as a promising economical and effective process to disperse the AlN NPs into Elektron21 (E21)  
7 alloy [4]. With this treatment, the AlN NPs could be incorporated into Mg matrix. The creep  
8 resistance of E21 with AlN NPs was improved. However, microstructural observations  
9 indicated some noticeable AlN clusters still existed in the matrix [11]. Moreover, the ultrasonic  
10 treatment is not suitable for industrial scale manufacturing. Therefore, there still exists a strong  
11 demand to search for a reliable and economical processing to fabricate the nanoparticle-  
12 reinforced Mg matrix composites (PMMCs). Recently, BCAST at Brunel university introduced  
13 a high shearing dispersion technology (HSDT) to stir the melt [12]. It can produce high energy  
14 dispersion rate to break up particle clusters and produce a better dispersion of particles. This  
15 technique was previously applied to disperse the SiC micro-sized particles in the AZ31 alloy  
16 [13]. It was found that with the assistance of HSDT the distribution of micro-sized SiC particles  
17 was quite satisfying.

18 In the current work, E21 is used as the base alloy, nano-sized AlN/Al NPs with the content  
19 of 0.25 wt.%, 0.5 wt.% and 1 wt.% were added for reinforcements. The creep properties of  
20 composites at 240 °C were investigated and the responsible mechanisms were explored and  
21 discussed.

## 22 2. Experimental procedures

23 E21 has a composition of Mg-2.85Nd-0.92Gd-0.41Zr-0.29Zn (all compositions are in wt.%  
24 hereafter unless specified) and was provided by Magnesium Elektron Ltd at Manchester, UK.  
25 Its composites were reinforced with different of AlN/Al NPs contents ranging from 0.25% to  
26 1% fabricated by HSDT. The AlN/Al NPs with an average size of 80 nm were synthesized with  
27 aluminum (Al) wires in an argon and nitrogen atmosphere. Al wires not only reacted with  
28 nitrogen to form AlN NPs, but were also retained as metallic Al [14]. The received NPs  
29 contained 75% AlN and 25% metallic Al (here termed as AlN/Al NPs). The HSDT with a  
30 rotor-stator device, developed by BCAST at Brunel University [12, 15], was applied to disperse  
31 the AlN/Al NPs clusters [13, 16]. The device was made of a  $\Phi$  80 mm open cylinder stator and  
32 a  $\Phi$  67.5 mm motor driven rotor with 4-bladed impeller. The rotation speed can be changed  
33 from 0-3000 rpm. The detailed mechanisms and procedure can be found elsewhere [15, 17, 18].

1 E21 with a weight of 12 kg was molten at 720 °C in a mild steel crucible with the protective  
2 gas (Ar+1 vol% SF<sub>6</sub>). The AlN/Al NPs with an amount of 0.25%, 0.5% and 1% were  
3 incorporated using HSDT at a rate of 3000 rpm for 1 min, respectively. After shearing, 3 kg of  
4 the molten melt was transferred into a three-zone resistance ring furnace at 670 °C for 5 min  
5 and was then immersed into a water bath at a speed of 100 mm/ min. For comparison, the as  
6 cast E21 without nanoparticle addition was also prepared.

7 The Vickers hardness measurements (HV5) were performed using an EMCO M1C 010  
8 machine with a 5 kg load and a dwell time of 30 s. At least 10 indentations for each condition  
9 were made. Compression creep tests of the specimens with a dimension of  $\Phi$  6 mm and 15 mm  
10 in length were carried out at 240 °C under a constant stress between 80 and 140 MPa. Applied  
11 Test Systems (ATS) lever arm creep test systems with two thermocouples were used to evaluate  
12 the creep properties. The creep strains were measured using an extensometer connected with  
13 the measuring unit. The sample temperature evolution during creep was monitored by two  
14 thermocouples inside the chamber with an accuracy of  $\pm 1$  °C. Creep experiments were  
15 interrupted once the steady-state creep stage was achieved.

16 The samples with a size of about 10×10×10 mm for microstructural observations were cut  
17 from cast cylinders. They were embedded in epoxy resin, which were observed along the  
18 longitudinal plane paralleled to the cooling direction. Before optical microscopy (OM)  
19 observation, samples were ground with silicon carbide abrasive paper from 500 grit to 2500  
20 grit and were polished with oxide polishing suspension (OPS). Then an acetic-picral solution  
21 with 8 g picric acid, 5 ml acetic acid, 10 ml distilled water and 100 ml ethanol was used for  
22 etching the samples. Metallographic characterization was performed using Leica DMI5000  
23 equipped with a digital camera and AnalySIS Pro software. Average grain size was evaluated  
24 using the linear intercept method [19] by AnalySIS Pro software. Line segments were drawn  
25 in the optical micrographs. The number of times each line segment intersects a grain boundary  
26 was counted. The average grain size was calculated as the ratio of intercept number to line  
27 length. The distribution of the grain size was determined to evaluate the uniformity degree. In  
28 order to gather quantitative grain size data, at least 50 counts were measured randomly in the  
29 optical micrographs. The number in one grain size range was divided by the total number of  
30 measured grains, and its value means the relative frequency in the whole size range. Scanning  
31 electron microscope (SEM) observations were performed using a TESCAN VEGA3-SB (Brno,  
32 Czech Republic) equipped with a Tescan energy dispersive X-ray spectrometer (EDS). Phase

1 identifications were performed on the high-energy X-ray beamtime HEMS P07 at PETRA III  
2 (DESY Hamburg) with wave length  $\lambda=0.01424$  nm.

### 3 3. Results

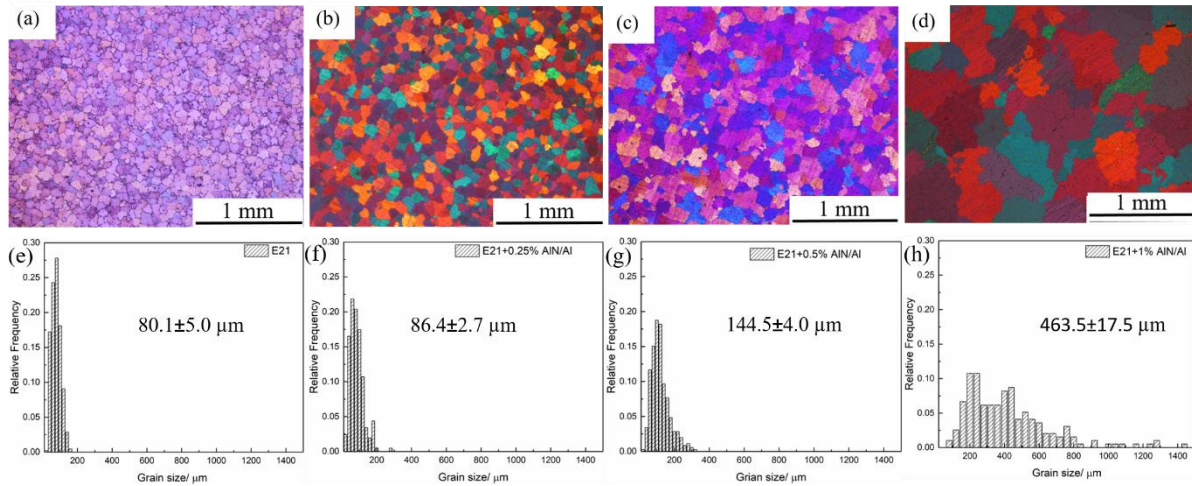
#### 4 3.1 Optical microstructure

5  
6 The casting parameters, grain sizes and hardness of E21 and E21 reinforced with AlN/Al NPs  
7 are listed in Table I. The hardness is not significantly influenced by the addition of AlN/Al NPs  
8 ranging from 0.25% to 0.5%. With the addition of AlN/Al NPs up to 1%, the hardness shows  
9 a noticeable increase to  $51.1\pm3.6$  compared with monolithic E21 alloy (Table I). Fig. 1 shows  
10 the microstructures of as cast E21 and E21 with the addition of 0.25%, 0.5%, and 1% AlN/Al  
11 NPs by HSDT. E21 exhibits equiaxed grains with an average grain size of  $80.1\pm5.0$   $\mu\text{m}$  (Fig.  
12 1(a)). With the addition of different contents of AlN/Al NPs, the grain size shows a linear  
13 increase from  $86.4\pm2.7$  to  $463.5\pm17.5$   $\mu\text{m}$  with increasing the content of AlN/Al NPs from  
14 0.25% to 1%. E21, E21+0.25% AlN/Al and E21+0.5% AlN/Al alloys have a range of grain  
15 size from 20-300  $\mu\text{m}$ . In contrast, E21+1% AlN/Al alloy has a wide range of grain size from  
16 20-1400  $\mu\text{m}$  (Fig. 1(e-h)). This indicates that the distribution of grain size is relatively uniform  
17 in the E21 with the addition of 0.25% and 0.5% AlN/Al NPs. By adding 1% AlN/Al NPs, the  
18 distribution of grain size becomes inhomogeneous, which mainly scattered in the range from  
19 200 to 600  $\mu\text{m}$ .

20 Table I. The casting parameters, hardness and grain size of E21 and E21 reinforced with  
21 AlN/Al NPs.

Matrix	AlN/Al [wt.%]	Stirring speed [rpm]	Stirring time [min]	Grain size [ $\mu\text{m}$ ]	Hardness [HV5]
E21	0	-	-	$80.1\pm5.0$	$45.0\pm1.2$
E21	0.25	3000	1	$86.4\pm2.7$	$46.3\pm2.6$
E21	0.5	3000	1	$144.5\pm4.0$	$45.7\pm2.4$
E21	1	3000	1	$463.5\pm17.5$	$51.1\pm3.6$

22



1

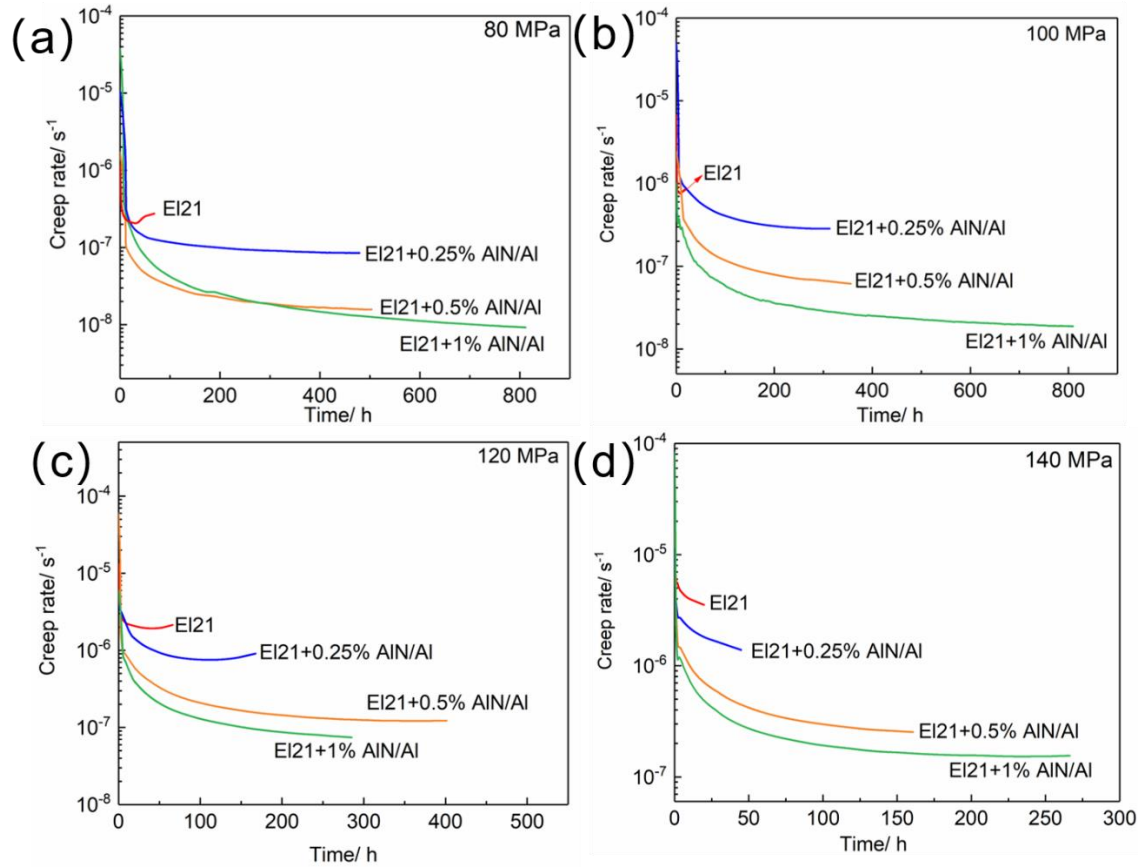
2 Fig. 1. Optical microstructures of as cast (a) E21, (b) E21+0.25% AlN/Al NPs, (c) E21+0.5%  
 3 AlN/Al NPs, (d) E21+1% AlN/Al NPs. The corresponding grain size distribution of (a-d) are  
 4 shown in (e-h), respectively.

### 5 3.2 Creep properties

6

7 Fig. 2 shows the typical curves of creep rate as a function of creep time under a stress  
 8 of 80, 100, 120 and 140 MPa at 240 °C. The creep rates decrease over time for the as cast  
 9 E21 and E21 composites. Table II shows the corresponding minimum creep rates of these  
 10 materials. It indicates that the minimum creep rate reduces apparently with the increasing  
 11 amount of AlN/Al NPs in E21 composites under all applied stress. For instance, the  
 12 minimum creep rate decreases from  $8.56 \times 10^{-8} \text{ s}^{-1}$  for E21+0.25% AlN/Al to  $9.05 \times 10^{-9} \text{ s}^{-1}$   
 13 for E21+1% AlN/Al NPs composite under a stress of 80 MPa. With the addition of 1%  
 14 AlN/Al NPs, the creep resistance was enhanced by about one order of magnitude.

15



1  
2 Fig. 2 Creep rate as a function of creep time at 240 °C (a) 80 MPa, (b) 100 MPa, (c) 120 MPa,  
3 (d) 140 MPa.

4  
5 Table II. Minimum creep rate under different applied stresses at 240 °C. The stress exponent  $n$   
6 was calculated by plotting the logarithmic stress vs. minimum creep rate.

	Stress [MPa]	E121	E121+0.25% NPs	E121+0.5% NPs	E121+1% NPs
$\dot{\epsilon}$ [s <sup>-1</sup> ]	80	$2.06 \times 10^{-7}$	$8.56 \times 10^{-8}$	$1.57 \times 10^{-8}$	$9.05 \times 10^{-9}$
	100	$6.87 \times 10^{-7}$	$2.77 \times 10^{-7}$	$6.14 \times 10^{-8}$	$1.87 \times 10^{-8}$
	120	$1.88 \times 10^{-6}$	$7.5 \times 10^{-7}$	$1.23 \times 10^{-7}$	$7.36 \times 10^{-8}$
	140	$3.68 \times 10^{-6}$	$1.47 \times 10^{-6}$	$2.53 \times 10^{-7}$	$1.52 \times 10^{-7}$
$n$		5.2	5.1	4.9	5.2

8  
9 Based on previous literatures [20-22], the relationship between minimum creep rate  $\dot{\epsilon}_s$  and  
10 applied stress  $\sigma$  can be described as Equation 1.

1 
$$\dot{\epsilon}_s = A\sigma^n \exp\left(-\frac{Q_c}{RT}\right) \quad (1)$$

2 A is a constant associated with the frequency factor and Burgers vector,  $\sigma$  is the applied stress,  
3  $Q_c$  is the activation energy during creep, R is the gas constant ( $8.314\text{J mol}^{-1}\cdot\text{K}^{-1}$ ). n is the stress  
4 exponent which implies the deformation mechanism responsible for creep. n values were  
5 calculated by the slope of  $\ln\dot{\epsilon}_s$  against  $\ln\sigma$ , as Equation 2 shows.

6 
$$\ln\dot{\epsilon}_s = \ln A + n\ln\sigma - \frac{Q_c}{RT} \quad (2)$$

7 The calculated stress exponent n values were listed in Table II. They vary from 4.9 to 5.2  
8 for these four alloys. Previous investigations indicated that n=5 is associated with dislocation  
9 climbing mechanism [23]. It can therefore be concluded the present dominant mechanism for  
10 E21 and its composites should be dislocation climbing-controlled creep.

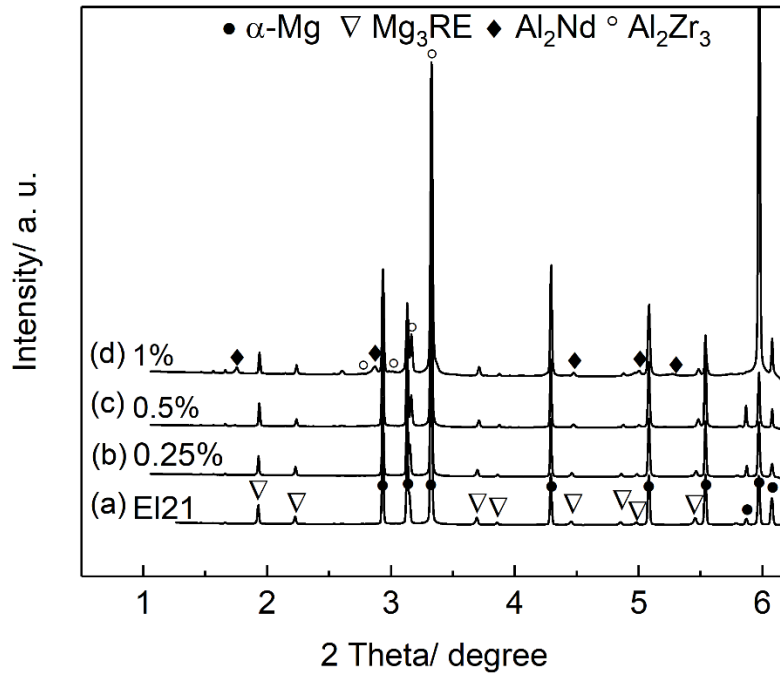
11 **3.3 Phase identification and SEM characterization**

12

13 Fig. 3(a) shows the high-energy X-ray diffraction (XRD) patterns of as cast E21, which is  
14 mainly composed of  $\alpha$ -Mg and  $\text{Mg}_3(\text{Nd, Gd})$  ( $\text{Mg}_3\text{RE}$ ) intermetallic compound. This result is  
15 in agreement with that reported by previous papers [11, 24, 25]. In this intermetallic compound  
16  $\text{Mg}_3(\text{Nd, Gd})$ , Nd was partially substituted by Gd without any change of the crystal structure.  
17 It is reported Nd ( $r=0.1821\text{ nm}$ ) has the similar atomic radii to Gd ( $r=0.1802\text{ nm}$ ) [11]. With  
18 the addition of 0.25% AlN/Al NPs in E21, it is difficult to detect any additional peaks besides  
19  $\alpha$ -Mg and  $\text{Mg}_3\text{RE}$  phases (Fig. 3(b)). With further additions of AlN/Al NPs up to 0.5% and 1%,  
20 some additional peaks with relatively low intensity caused by the phases  $\text{Al}_2\text{Nd}$  and  $\text{Al}_2\text{Zr}_3$   
21 were identified (Fig. 3(c) and (d)).

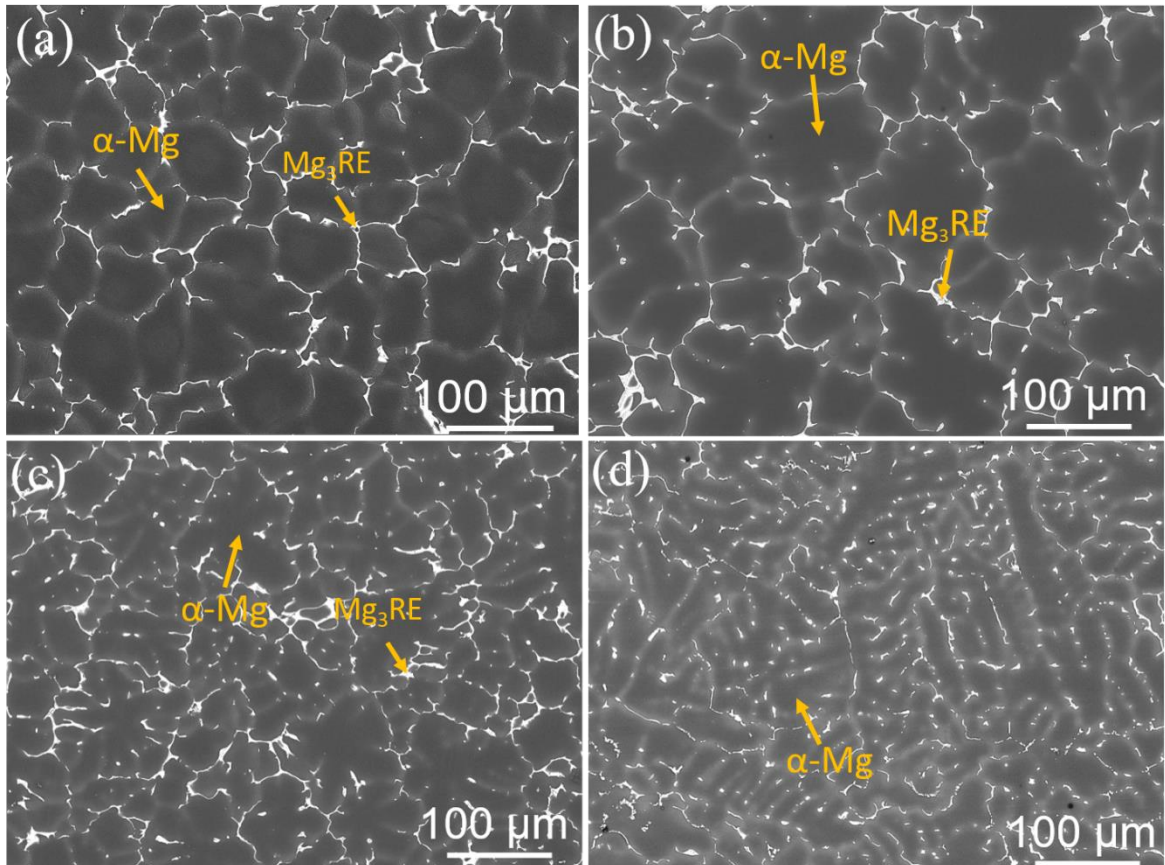
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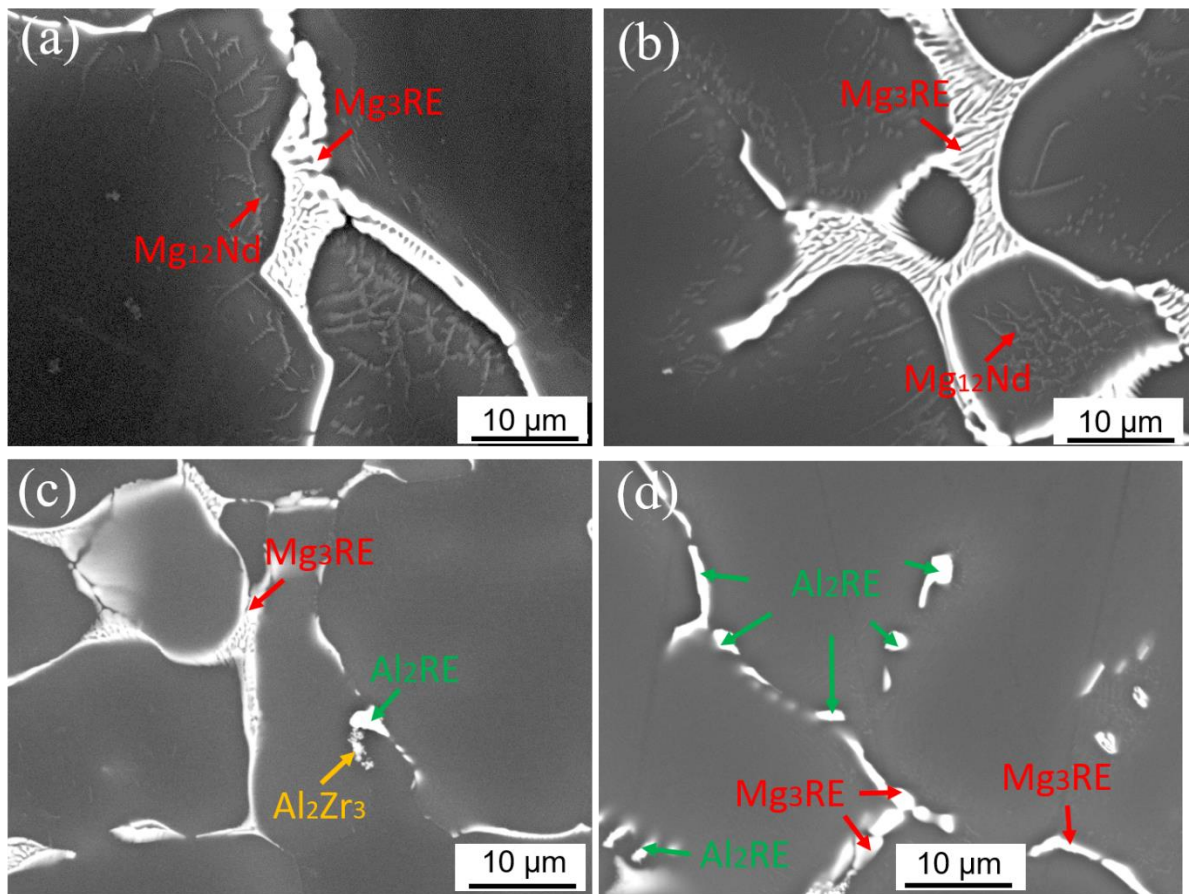
2 Fig. 3 High-energy XRD patterns of the as cast (a) E21 alloy, (b) E21+0.25% AlN/AI NPs and  
3 (c) E21+0.5% AlN/AI NPs and (d) E21+1% AlN/AI NPs.



4

1 Fig. 4 Back scattered electron (BSE) SEM images of as cast (a) E21, (b) E21+0.25% AlN/Al  
2 NPs, (c) E21+0.5% AlN/Al NPs and (d) E21+1% AlN/Al NPs.

3 Fig. 4(a-c) shows the microstructures of as cast E21, E21+0.25% AlN/Al NPs and E21+0.5%  
4 AlN/Al NPs with  $\alpha$ -Mg and  $Mg_3RE$  (bright phase). With the addition of AlN/Al NPs up to 1%,  
5 according to high-energy XRD results, the intermetallics are mainly composed of  $Mg_3RE$ ,  
6  $Al_2Zr_3$  and  $Al_2RE$  phase. Without the AlN/Al NPs in E21,  $Mg_3RE$  phases are normally  
7 distributed along the dendritic boundaries (Fig. 4(a)). By adding AlN/Al NPs (Fig. 4(b-d)), the  
8 distribution of the eutectic becomes discontinuous and more homogeneous in the matrix  
9 compared with the unreinforced E21. Especially, with the addition of AlN/Al NPs up to 1%,  
10 the eutectic becomes much finer, denser and hyper-dendritic forming in the matrix.



12 Fig. 5 BSE micrographs of as cast (a) E21, (b) E21+0.25% AlN/Al NPs, (c) E21+0.5%  
13 AlN/Al NPs and (d) E21+1% AlN/Al NPs.

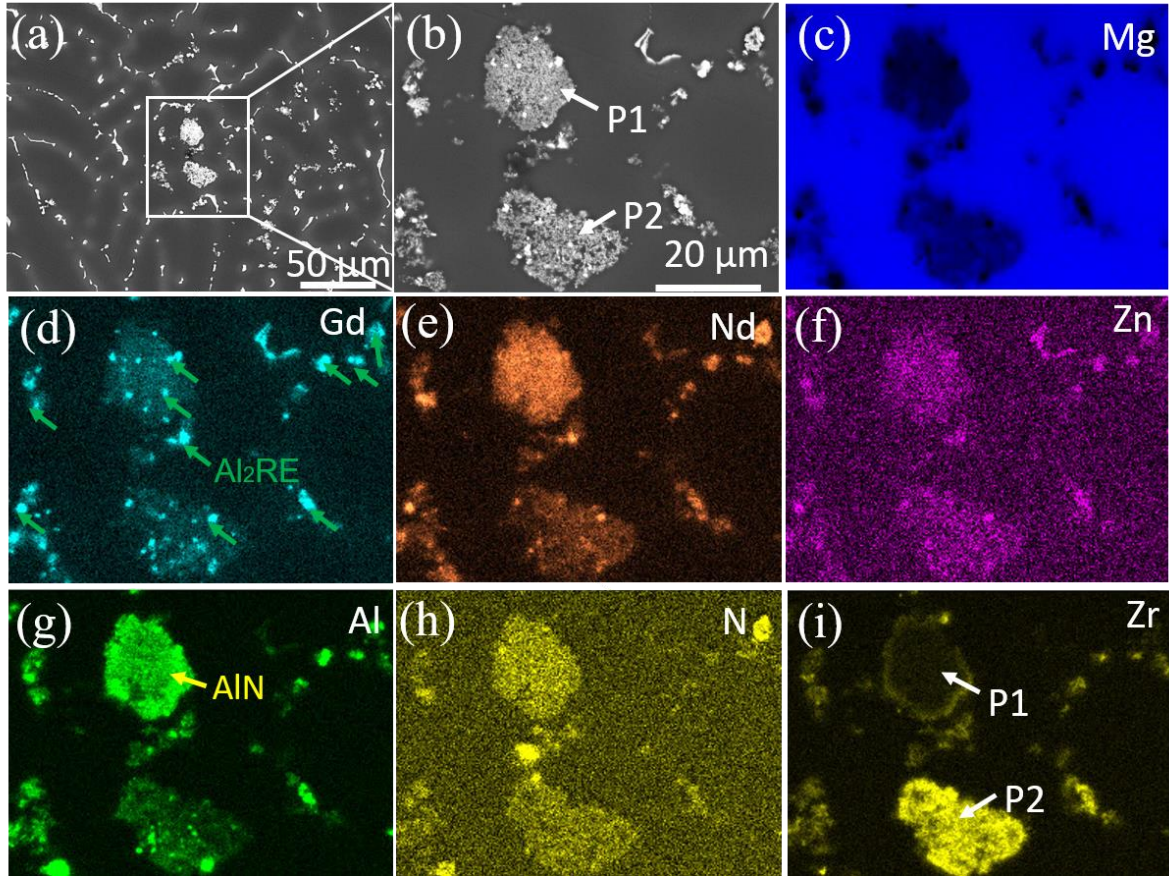
14 Fig. 5(a) exhibits the lamellar  $Mg_3RE$  intermetallic phase and some lath-like precipitates  
15 located around the bright phase. These precipitates were reported to be  $Mg_{12}Nd$  (tetragonal  
16 structure), which normally exist in Mg-Nd alloys due to the supersaturated of Nd [26, 27]. A

1 similar phenomenon was also observed in the present E21+0.25% AlN/Al NPs composites (Fig.  
2 5(b)). After adding more AlN/Al NPs with 0.5% or 1%, the intermetallics were much refined  
3 than that of E21. Meanwhile, the Mg<sub>12</sub>Nd phase was difficult to be detected around the eutectic  
4 (Fig. 5(c) and (d)). This demonstrates that the addition of AlN/Al NPs influences the formation  
5 of Mg<sub>12</sub>Nd. With increasing the content of AlN/Al NPs from 0.5% to 1%, an additional phase  
6 Al<sub>2</sub>RE was observed (Fig. 5(c-d)). With the assistance of HSDT, it is difficult to observe any  
7 agglomerations of AlN/Al NPs in E21 with the additions of 0.25% and 0.5% AlN/Al NPs,  
8 indicating a relative homogeneous distribution of AlN/Al NPs. However, when the content of  
9 AlN/Al NPs is up to 1%, some clusters are still visible such as at P1 and P2 (Fig. 6(a)).  
10 According to the EDS mapping, P1 was confirmed as AlN NPs (Fig. 6(g) and (h)). Although  
11 the morphologies of P1 and P2 are similar (Fig. 6(b)), their contents of Zr are different. At P1  
12 Zr element was only found in the outer layer of AlN NPs (Fig. 6(i)), implying that Zr only  
13 reacted with AlN NPs at their surfaces. Compared with P1, P2 has a much higher content of  
14 Zr. EDS point analysis also shows that besides Mg, the cluster P2 is mainly composed of Al,  
15 N and Zr (Fig. 7). All these phenomena confirm the reaction of AlN/Al NPs with Zr has  
16 happened during casting and solidification. This conclusion can further be supported by  
17 previous results that that Zr could react with AlN to form aluminides Al<sub>2</sub>Zr and Al<sub>3</sub>Zr at a  
18 certain temperature [28]. Besides these two aluminides, it is reported that other aluminides  
19 could also be formed. S. N. Agafonov et al. [29] reported that Al reacted with Zr to form a  
20 different stable Al<sub>2</sub>Zr<sub>3</sub> phase. Based on the present results of high-energy XRD, it can be  
21 concluded that at P2 the phase Al<sub>2</sub>Zr<sub>3</sub> was formed .

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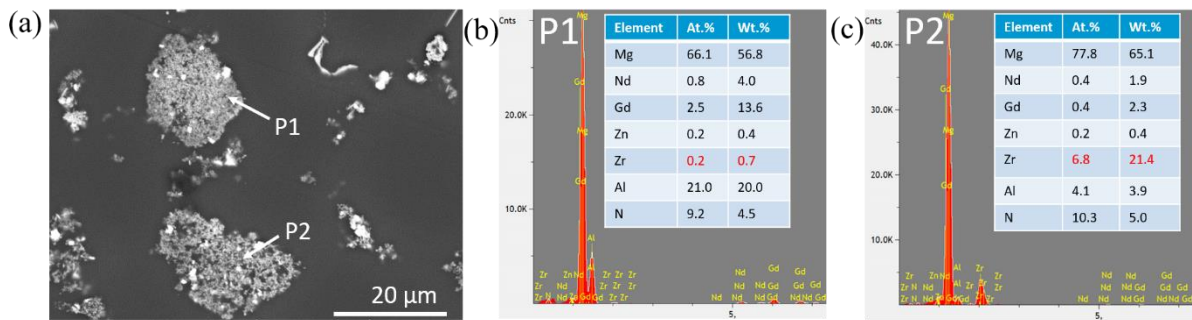
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Fig. 6. (a) BSE micrograph of E21+1% AlN/Al NPs composite and (b) the magnified image of (a), (c-i) EDS mapping analysis of E21+1% AlN/Al NPs composite.

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Fig. 7 BSE micrograph of E21+1% AlN/Al NPs composite showing the clusters of P1 and P2 and (b-c) the EDS point detection of P1 and P2 in (a).

8

## 4. Discussion

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AlN NPs were usually considered as effective heterogeneous nucleation sites for the formation of  $\alpha$ -Mg in Mg-based alloys [30]. HSdT was demonstrated to be an effective approach to refine the grain size in Mg alloys [15]. However, in the present investigation, the grain coarsening was observed with the addition of AlN/Al NPs ranging from 0.25% to 1% in

1 E21 using HSDT. It should be related to the poisoning mechanism of Al element in Zr-  
2 containing E21. The grain refiner Zr reacted with Al as well as AlN NPs to form the stable  
3 intermediate Al-Zr phase. Consequently, the efficiency of the grain refinement was  
4 significantly weakened by the loss of Zr. With the assistance of HSDT, the melt was intensively  
5 stirred and the possibility of contact between AlN/Al and Zr was increased. A greater amount  
6 of Zr was consequently consumed during solidification and then the grain became much larger.  
7 The identification of Al<sub>2</sub>Zr<sub>3</sub> phases in the composites with the AlN/Al contents up to 0.5% and  
8 1% further proves the loss of grain refiner Zr.

9 The main deformation mechanisms responsible for the second steady-state creep include  
10 dislocation creep, diffusion creep and grain boundary sliding creep [23, 31, 32]. According to  
11 the calculation of stress exponent n, it is concluded that the controlling mechanism in the  
12 AlN/Al NPs reinforced E21 composites is dislocation climbing creep. In the present  
13 investigation, the additional effects from the change of grain size on creep resistance could be  
14 ignored, since the responsible creep mechanism is dislocation climbing rather than the grain  
15 boundary sliding. This is due to the fact that at grain boundaries there exist lots of intermetallic  
16 particles such as Mg<sub>3</sub>RE. The phases/precipitates which can hinder the dislocation movement  
17 and pin the grain/dendritic boundary sliding should be beneficial for the improvement of creep  
18 resistance at elevated temperature. In the AlN/Al NPs reinforced E21 composites, Mg<sub>3</sub>RE  
19 phase is the main intermetallic compound, which is a thermal stable phase at the creep  
20 temperature 240 °C. It can act as an effective barrier to obstruct the dislocation movement and  
21 boundary sliding during the creep deformation. With further addition of AlN/Al NPs, the  
22 distribution of intermetallics phase becomes much more homogeneous and their size decreases.  
23 Such homogeneous and fine intermetallics can prevent the dislocation slip more efficiently  
24 during creep. Consequently, the creep resistance is improved.

25 The EDS mapping analysis of E21+1% AlN/Al NPs composite shows the presence of AlN  
26 clusters (Fig. 6(g)), indicating that not all the AlN NPs were consumed by the Zr. Part of AlN  
27 NPs still retained as ceramic particles in the matrix. Due to its small amount, no AlN peaks  
28 could be observed in the high-energy XRD patterns. According to Orowan strengthening [33],  
29 these remained NPs can hinder the dislocation movement, especially for these particles with  
30 small size and relative homogeneous distribution. Therefore, the AlN NPs can contribute to the  
31 improvement of creep resistance to some extent.

1 With increasing the content of AlN/Al NPs from 0.5% to 1%, the amount of Al<sub>2</sub>RE also  
2 increased (Fig. 5(c-d)). They are interconnected with Mg<sub>3</sub>RE to occupy the dendritic  
3 boundaries in the matrix. Amberger et al. [34] investigated MRI230D alloy and found that the  
4 highly connected secondary phase Al<sub>2</sub>Ca along the boundaries enhances the creep resistance  
5 because they can effectively shield load from the soft  $\alpha$ -Mg matrix. It is believed that the  
6 formation of Al<sub>2</sub>RE caused by the reaction of Al with REs also contribute to the improvement  
7 of creep resistance for the AlN/Al NPs reinforced E21 composites. Normally, the thermal  
8 stability of intermetallics is proportional to their melting points, the higher their melting point  
9 the higher their thermal stability. The melting point of Al<sub>2</sub>RE is 1200 °C which is much higher  
10 than that of Mg<sub>3</sub>RE (780 °C) [35], implying it has a better thermal stability than Mg<sub>3</sub>RE. Zhu  
11 et al. [36] also reported that Al<sub>2</sub>RE was stable at temperatures up to 200 °C without any distinct  
12 decomposition. Al<sub>2</sub>RE may provide positive effects on the improvement of creep resistance at  
13 elevated temperatures to some extent. Moreover, although its amount is small, the formation  
14 of Al<sub>2</sub>Zr<sub>3</sub> phase may be also beneficial for the improvement of creep properties. Kabirian et al  
15 [37] reported that the creep resistance of AZ91 with the addition of Zr in the range of 0.2-1.0  
16 wt.% were enhanced significantly due to the occurrence of Al-Zr intermetallics. Al-Zr  
17 intermetallic particles could act as thermal stable particles to oppose the revery process. In  
18 summary, the increase in the fraction of the intermetallic phases as a result of the reaction of  
19 AlN/Al NPs with the alloying elements RE and Zr that is helpful for the improvement in creep  
20 resistance.

## 21 Conclusions

22 The E21 composites reinforced with 0.25%, 0.5% and 1% AlN/Al NPs reinforced E21  
23 composites were successfully prepared by HSDT. The following conclusions can be drawn:

24 1. The grain size exhibits an obvious increase in AlN/Al NPs reinforced E21  
25 composites compared with monolithic E21. With increasing content of the AlN/Al NPs,  
26 the grain becomes larger. This is attributed to the poison mechanism of Al addition in this  
27 Zr-containing E21.

28 2. The creep resistance is improved for the E21 with the addition of AlN/Al NPs.  
29 With the addition of 1% AlN/Al NPs, the creep resistance increases by more than one order  
30 of magnitude. The responsible strengthening mechanism is mainly ascribed to the  
31 homogeneous and refined intermetallics at dendritic boundaries. They can inhibit  
32 dislocation movement more effectively during creep.

1           3.       The formation of additional Al<sub>2</sub>RE and Al<sub>2</sub>Zr<sub>3</sub> phases by the reaction of AlN/Al  
2       NPs and the alloying elements REs and Zr can contribute to the improvement of creep  
3       resistance. Meanwhile, the remained AlN NPs without reaction are also beneficial for  
4       improving the creep resistance due to Orowan strengthening.

#### 6 Acknowledgement

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